

Okayama University Medical Research Updates (OU-MRU)

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Okayama University research: High-quality growth

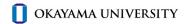
(Okayama, 9 September) Researchers at Okayama University report in ACS Nano how highly crystalline samples of transition metal dichalcogenides such as MoS₂ and WS₂ can be grown. These ultra-thin and mechanically flexible materials are important for the development of flexible optoelectronic device applications.

Transition metal dichalcogenides (TMDCs) are a class of materials with physical properties that make them ideally suited for use in flexible optoelectronic applications, such as light detectors, light-emitting diodes and solar cells. For such applications to perform well, the crystalline quality of the TMDCs needs to be extremely high, however; defects in the crystal structure worsen device performance. The crystalline quality of a sample is related to the number of grain boundaries — interfaces between different grains, or domains, within the crystal. Different domains have the same chemical composition and structure but are oriented differently with respect to each other. The lower the number of grain boundaries, the larger the domains, and the better the sample's crystallinity. Now, Assistant Professor SUZUKI Hiroo and HASHIMOTO Ryoki (graduate students) Okayama University and colleagues have developed a technique with which highly crystalline TMDCs can be grown. Furthermore, the technique enables to optimize the performance of TMDCs for optoelectronic devices.

The researchers' approach is based on chemical vapor deposition (CVD), a method in which a substrate is put in a vacuum chamber and is exposed to particular chemical vapors. This leads to chemical reactions and depositions on the substrate, resulting in the growth of the desired material. Assistant Professor SUZUKI and colleagues grew the TMDCs MoS₂ (molybdenum disulfide) and WS₂ (tungsten disulfide), for which they let metal salts and sulfur sources react in the CVD chamber. What was special is that they used a stacked substrate configuration: two silicon-based substrates put close to each other, creating a confined environment for the TMDCs to form. This type of 'microreactor' led to the growth of samples with large domains.

Through careful analysis of the obtained crystals' morphology and numerical modelling of the processes at play, Assistant Professor SUZUKI and colleagues deduced that the growth is governed by surface diffusion, that is, by atoms moving along the substrate surface. This leads to a lower probability for atoms to nucleate and initiate the growth of a domain; instead of several small domains, only one or a few large domains develop.

The scientists then investigated how the growth temperature used in the CVD process influenced the materials' photoluminescence — the ability to emit light of a particular wavelength after being irradiated with light of another wavelength. They found that for a growth temperature of about 820°C, the photoluminescence characteristics of WS₂ are optimal.



In conclusion, the reported growth method produces, in a controlled way, highly crystalline TMDC samples that are suitable for use in optoelectronic applications. Quoting the researchers: "These findings should significantly contribute to the realization of applications of high-performance optoelectronic devices based on high-quality monolayer TMDCs."

Background

Chemical vapor deposition

In chemical vapor deposition (CVD), one or more gases are made to pass over a hot substrate's surface under (low) vacuum. The gases dissociate on the substrate or in its immediate vicinity, after which the decomposition products accumulate on the surface. Often, by-products are formed in the process; these are removed from the reaction chamber through gas flow.

By choosing the gases that are injected into the CVD chamber, specific materials can be grown. Various types of materials structures can be obtained, including monocrystalline (consisting of one domain only), polycrystalline (consisting of several domains) or amorphous materials.

Assistant Professor SUZUKI Hiroo and HASHIMOTO Ryoki from Okayama University and colleagues have now used a CVD approach to grow highly crystalline samples of transition metal dichalcogenides, specifically MoS₂ and WS₂, by using a stacked substrate setup; the resulting confinement provides optimal conditions for monocrystals to grow.

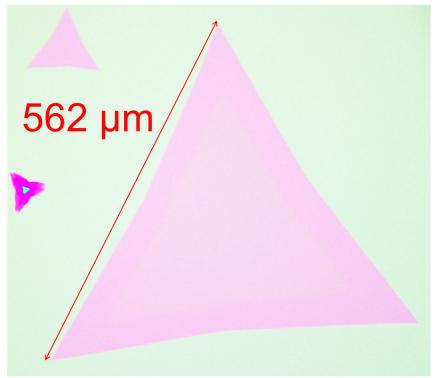


Figure
Optical micrograph of large size monolayer WS₂ crystal.



Reference

Hiroo Suzuki, Ryoki Hashimoto, Masaaki Misawa, Yijun Liu, Misaki Kishibuchi, Kentaro Ishimura, Kenji Tsuruta, Yasumitsu Miyata, and Yasuhiko Hayashi. Surface Diffusion-Limited Growth of Large and High-Quality Monolayer Transition Metal Dichalcogenides in Confined Space of Microreactor. *ACS Nano*, 2022, 16, 11360–11373.

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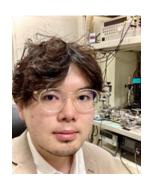
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The whole volume: OU-MRU (1-)

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- Vol.5: Cell injections get to the heart of congenital defects
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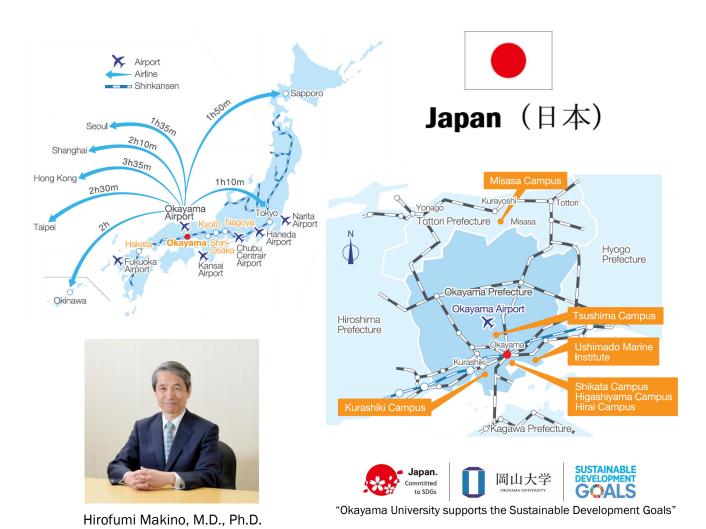


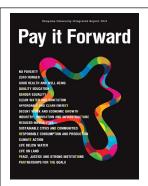
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Okayama University is one of the largest comprehensive universities in Japan with roots going back to the Medical Training Place sponsored by the Lord of Okayama and established in 1870. Now with 1,300 faculty and 13,000 students, the University offers courses in specialties ranging from medicine and pharmacy to humanities and physical sciences.

Okayama University is located in the heart of Japan approximately 3 hours west of Tokyo by Shinkansen.

Website: http://www.okayama-u.ac.jp/index_e.html





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An integrated report is intended to explain how an organization creates value over time through an organic integration of the vision and the combination of financial information and other information. Through this report we hope to promote greater interest in Okayama University among readers everywhere. In order to help us make improvements in future editions, we encourage you to contact us with any comments and suggestions you may have.

President, Okayama University